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Roasted coffee wastes as a substrate for *Escherichia coli* to grow and produce hydrogen

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One sentence summary: Roasted coffee wastes can be applied for biomass and biohydrogen production by *Escherichia coli*. Responsible hydrogenase enzymes and enhanced hydrogen production have been detected.

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ABSTRACT

After brewing roasted coffee, spent coffee grounds (SCGs) are generated being one of the daily wastes emerging in dominant countries with high rate and big quantity. *Escherichia coli* BW25113 wild-type strain, mutants with defects in hydrogen (H₂)-producing/oxidizing four hydrogenases (Hyd) ($\Delta hyaB \Delta hycB$, $\Delta hycE$, $\Delta hycF$) and septuple mutant ($\Delta hyaB \Delta hycB \Delta hycC \Delta hycE \Delta hycF \Delta hycG \Delta hycH$) were investigated by measuring change of external pH, bacterial growth and H₂ production during the utilization of SCG hydrolysate. In wild type, H₂ was produced with rate of 1.28 mL H₂ (g sugar)⁻¹ h⁻¹ yielding 30.7 mL H₂ (g sugar)⁻¹ or 2.75 L (kg SCG)⁻¹ during 24 h. In septuple mutant, H₂ production yield was 72 mL H₂ (g sugar)⁻¹ with rate of 3 mL H₂ (g sugar)⁻¹ h⁻¹. H₂ generation was absent in *hycE* single mutant showing the main role of Hyd-3 in H₂ production. During utilization of SCG wild type, specific growth rate was 0.72 ± 0.01 h⁻¹ with biomass yield of 0.3 g L⁻¹. Genetic modifications and control of external parameters during growth could lead to prolonged and enhanced microbiological H₂ production by organic wastes, which will aid more efficiently global sustainable energy needs resulting in diversification of mobile and fixed energy sources.

Keywords: roasted coffee wastes; waste pretreatment; hydrogenases; biomass and bio-H₂ production

INTRODUCTION

Coffee is the most preferred drink worldwide and by its consumption rebates only petroleum (Murthy and Naidu 2012). It is grown in more than 70 countries and produces to over 16 billion pounds of coffee beans every year (Blinova et al. 2017), which is consumed as a soluble or instant coffee production, accompanied with by-products generation. While using them once and throwing, a huge amount of waste is generated and the

presence of organic material is pollutant leftovers and furnishes a part of toxic nature (Campos-Vega et al. 2015).

Spent coffee grounds (SCGs) are the main coffee industry residues, with coffee silverskin (CS), obtained from different stages of coffee bean treatment (Mussatto et al. 2011a). SCG residue is generated from both soluble and instant coffee making, and waste quantity reaches over 6 million tons per year (Kovalcik, Obrucaa and Marova 2018). It is worth to note that ~650 kg of SCGs are generated from 1 ton green coffee, and ~2 kg

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from 1 kg of soluble coffee is produced (Murthy and Naidu 2012). After brewing, roasted coffee wastes (the solid residue emerged during the treatment of coffee with hot water or steam) can be considered as SCG (Getachew and Chun 2017). Its hydrolysate contains fermentable sugars, such as glucose, galactose, mannose, arabinose and cellobiose, which are polymerized into lignocellulose structures (cellulose 5–12 wt%, hemicellulose 40 wt%). It is richer than CS-containing material. Also proteins and fat constitute a significant fraction of SCGs (Scully, Jaiswal and Abu-Ghannam 2016) and have great applications in different scope of life starting from cosmetics to wastewater treatment possibilities, or as nutritional supplement in animal feed (Kovalcik, Obruca and Marova 2018), but still making some value from SCGs is needed, so studies are ongoing (Mussatto et al. 2011a). Besides the organic compounds, a variety of mineral elements are present in SCGs and CS ashes (Mussatto et al. 2011b).

It has been shown that raw materials or wastes can be applied to generate biogas, bioethanol, biomethane or other valuable chemicals (Batista et al. 2016). Besides methane, bio-H₂ can also be produced from various industrial or agricultural organic wastes (Poladyan et al. 2018; Prabakar et al. 2018). Biological H₂ production is found to be more environmentally friendly, renewable and less energy consuming compared to water electrolysis, natural gas reforming and coal gasification (Lee, Vermaas and Rittmann 2010; Trchounian, Sawers and Trchounian 2017). H₂ is used widely in fuel cells to generate electricity (Rahman et al. 2016), which results in decreasing greenhouse gases, and by increasing the use of raw material (organic waste) the positive impact for environment will be multiplied. H₂ can be produced by well-studied *Escherichia coli* via utilization of pretreated brewery spent grains (BSG) or distiller's grains (DG) (Sargsyan et al. 2016; Poladyan et al. 2018). In addition, glycerol is generated as a waste product during biodiesel production, which made it much more inexpensive in recent years (Clomburg and Gonzalez 2013; Trchounian and Trchounian 2015) and could be used solely or in mixture with various wastes as a feedstock for biomass and bioenergy or other products with added values (Trchounian, Sawers and Trchounian 2017). Interestingly, *E. coli* can utilize mixture of sugars and glycerol simultaneously (Wang et al. 2019). All this is important for applying in bio-based circular economy to control the application and usage of various organic wastes (D'Amato et al. 2017).

Escherichia coli performs mixed-acid fermentation and can convert sugars (glucose, arabinose, galactose, xylose etc.) or alcohols (glycerol) into different organic acids (e.g. lactic, succinic, acetic, formic acids), ethanol, H₂ and CO₂ (Trchounian, Sawers and Trchounian 2017; Zhu, San and Bennett 2020). In lower stages of glycolysis, H₂ is generated from formate via reversible membrane-bound hydrogenases (Hyd), which have been extensively studied (Hollinshead et al. 2016; Sargent 2016; Benoit et al. 2020). *Escherichia coli* has four hydrogenase (Hyd) enzymes (Hyd-1 to 4), where Hyd-1 (*hya*) induced in the presence of formate in stationary growth phase is responsible for pH switch from alkaline to acidic values (Trchounian et al. 2011, 2012), and Hyd-2 is active in exponential growth phase in anaerobic conditions and also responsible for H₂ production during glycerol fermentation at slightly alkaline pH (Trchounian and Trchounian 2009, 2015; Trchounian, Sawers and Trchounian 2017). Hyd-3 (*hyc*) is mainly responsible for H₂ production during sugar utilization (Maeda, Sanchez-Torres and Wood 2007a; Redwood et al. 2008; Petrosyan et al. 2020). H₂ is generated via Hyd-3 and Hyd-4 (*hyf*) forming formate hydrogen lyase (FHL) complexes with formate dehydrogenase H (FDH-H), but

it depends on external pH and redox potential values (Mnat-sakanyan, Bagramyan and Trchounian 2004; Trchounian 2015; Trchounian, Sawers and Trchounian 2017). By manipulating *E. coli* genes encoding different components of FHL complex, it was possible to enhance H₂ production yield from formate by ~2.8-fold with high *E. coli* cell density of 93 g L⁻¹ with productivity of 300 L H₂ h⁻¹ at 37°C in continuous reactor (Yoshida et al. 2005). Moreover, it has been shown that by introducing multiple mutations, it was possible to create septuple mutant (BW25113 Δ *hyaB* Δ *hyc* Δ *hycA* Δ *fdoG* Δ *ldhA* Δ *frdC* Δ *aceE*; Table 1) for enhanced H₂ production from glucose. The strategy that has been applied was to redirect glucose mainly to formate generation that can be further oxidized by FHL complex to H₂ and CO₂ (Maeda, Sanchez-Torres and Wood 2007b).

Current paper describes the possibility of using SCG hydrolysate solely or in the mixture with glycerol as a feedstock for bio-H₂ production by *E. coli* and generation of bacterial biomass. Moreover, Hyd enzymes responsible for H₂ evolution have been identified, and enhanced H₂ production has been obtained by applying appropriate genetically engineered mutant. This is significant for developing bio-based technologies for application of available various organic wastes and simultaneous generation of two different beneficial products—biomass and by-products such as H₂ (Ghimire et al. 2015; Maurya, Singla and Negi 2015).

MATERIALS AND METHODS

Bacterial strains, cell specific growth rate and yield, H₂ production determination

Escherichia coli wild type (wt) BW25113, Hyd mutants with defects in different Hyds and septuple mutant were used (see Table 1).

Bacteria were cultivated in anaerobic conditions in 500-mL glass bottles with pretreated SCGs growth medium. 3% (v/v) of night culture was added to same concentrations of growth medium. Bacterial optical density at 600-nm wavelength (OD₆₀₀) was measured by spectrophotometer (UV-VIS spectrophotometer, Cary 60, Agilent Technologies, USA) and biomass yield was estimated by determining the dry weight of culture and expressed in g L⁻¹ (Poladyan et al. 2018). Bacterial specific growth rate (μ) was calculated as $\ln(\text{OD}_{600}) \text{ h}^{-1}$ taking the doubling values of exponential phase. Redox potential (ORP in mV) was determined by ORP electrode Pt BNC (HI3131, HANNA Instruments, Portugal), and pH by selective pH meter (HI1131, HANNA Instruments, Romania), as before (Trchounian et al. 2011). ORP electrode was calibrated by 16.13 g L⁻¹ K₃[Fe(CN)₆] and 21.12 g L⁻¹ K₄[Fe(CN)₆] (pH 6.86) reference solution, where the ORP was 254 ± 10 mV at 25°C (Trchounian and Trchounian 2009). Pt electrode is sensitive to H₂, and drop of ORP was detected during the utilization of substrate, and mineral composition does not affect ORP change and addition of H₂ does not change the external pH value (Piskarev et al. 2010). When ORP drops to strictly negative values (≥ -400 mV), H₂ production is determined. This method is similar to Clark-type electrode, which was shown by Noguchi et al. (2010).

Cumulative H₂ production measurements were performed in 500-mL glass vessels under permanently stirring conditions. The gases bubbled (H₂ and CO₂) were treated by 1 M NaOH solution (to eliminate CO₂ from gas mixture) and H₂ gas was collected and estimated by the water displacement volume (Poladyan et al. 2018). H₂ production rate was expressed as mL H₂ per g sugar utilized per h [mL H₂ (g sugar)⁻¹ h⁻¹] and H₂ yield—in mL per g

Table 1. Characteristics of bacterial strains used.

Strains	Genotype	Subunits lacking	Reference
BW25113	<i>lacIq rrmBT₁₄ ΔlacZ_{W116} hsdR514 ΔaraBAD_{AH33} Δrha BAD_{LD78}</i>	WT parental strain	Maeda, Sanchez-Torres and Wood 2007b
^a MW1000	BW25113 Δ <i>hyaB</i> Δ <i>hybC</i>	Large subunits of Hyd-1, Hyd-2	Maeda, Sanchez-Torres and Wood 2007b
^a JW2917	BW25113 Δ <i>hycE</i>	Large subunit of Hyd-3	Baba et al. 2006
^a JW2472	BW25113 Δ <i>hyfG</i>	Large subunit of Hyd-4	Trchounian et al. 2011
^a BW25113 <i>hyaB hybC hycA fdog ldhA frdC aceE</i>	BW25113 Δ <i>hyaB</i> Δ <i>hybC</i> Δ <i>hycA</i> Δ <i>fdog</i> Δ <i>ldhA</i> Δ <i>frdC</i> Δ <i>aceE</i>	Large subunit of Hyd-1 and 2, repressor of FHL, α -subunit of formate dehydrogenase-N, lactate dehydrogenase, fumarate reductase, pyruvate dehydrogenase	Maeda, Sanchez-Torres and Wood 2007b

^aResistant to kanamycin.

sugar utilized [mL H₂ (g sugar)⁻¹]. H₂ yield was also calculated as L H₂ per kg SCG used and expressed as L H₂ (kg SCG)⁻¹.

SCG treatment and determination of total carbohydrates

SCGs generated from coffee preparation containing mainly Robusta with a little amount of Arabica have been obtained from Zanussi Necta Zenith (Italy) coffee making machines in Yerevan, Armenia. Afterward, SCGs (with slightly acidity and humidity of 70%) have been dried at 100°C till constant mass observed and suspended in distilled water. Mixture was exposed to pretreatment with acidic hydrolysis in an autoclave (WiseClave WACS-1100, Daihan Scientific, South Korea) at 121°C for 45 min (Fig. 1). Various concentrations of SCG (40–100 g L⁻¹) were tested, and optimal was detected to be 40 g L⁻¹. For pretreatment, 180 mg H₂SO₄ per g SCGs was applied (Kovalcik, Obruca and Marova 2018; Poladyan et al. 2018; Cesaro et al. 2020). The carbohydrate (sugars) content was determined with the method according to Dubois et al. (1956).

The pH of the SCG hydrolysate was adjusted to pH 7.0 by addition of 24 g L⁻¹ K₂HPO₄. 13.6 g L⁻¹ glycerol was added as a substrate in the SCG medium when indicated.

Reagents and data processing

Glycerol, K₂HPO₄ and H₂SO₄ (98%) and other reagents with analytical grade were used. The average data was figured by at least three experiments performed, the standard errors were considered, and Student criteria (P) were calculated to confirm the difference in average data between experiments performed; the difference was considered valid when P < 0.05 (Sargsyan et al. 2016; Poladyan et al. 2018).

RESULTS AND DISCUSSION

Escherichia coli growth during utilization of SCG

It is well known that *E. coli* can grow and perform mixed-acid fermentation on various organic carbon sources that depend on diverse environmental conditions such as pH, concentration of carbon source, ORP etc. *Escherichia coli* wt and all mutants tested were able to utilize SCGs and generate biomass. The growth and H₂ production by *E. coli* in SCG medium were studied till 96 h in batch cultures.

During assays, higher H₂SO₄ concentrations were found to inhibit both bacterial growth (Fig. 2) and H₂ production. Moreover, when SCG medium pH was adjusted by KOH, bacteria did not grow >6 h because of low pH (pH 4.9) generated (data not shown), and thus for further experiments K₂HPO₄ was used. Also 1, 2 and 5 times SCG hydrolysate dilutions were investigated, and 2 times dilution was revealed to be optimal (data not shown). The optimal conditions for both bacterial growth and H₂ production were found to be 40 g L⁻¹ SCGs treated with 180 mg H₂SO₄ per g SCGs hydrolyzed in 121°C for 45 min, pH adjusted by 24 g L⁻¹ K₂HPO₄ and final medium diluted twice (Fig. 1). Note that many pretreatment methods and technologies of organic wastes exist, and depending on the method chosen the yield of end product will be different (Kumari and Singh 2018).

Wt specific growth rate (μ) during SCG and glycerol co-fermentation was 0.68 ± 0.02 h⁻¹ compared to that without glycerol where μ was 0.72 ± 0.01 h⁻¹ (Fig. 3). Indeed, bacteria distinguish between the substrates to utilize at first due to activation of appropriate genes and metabolic networks (Wang et al. 2019). *hyaB hybC* double mutant demonstrated similar μ as in wt, suggesting that Hyd-1 and Hyd-2 role for bacterial growth was absent in the conditions used. This was in good conformity with previously obtained data from different groups for glucose fermentative conditions (King and Przybyla 1999; Pinske et al. 2012). However, *hyfG* and septuple mutants showed significantly lower μ -2.8 and 2-fold decrease, respectively, in both conditions with and without glycerol, compared to wt (see Fig. 3). In *hycE*, single mutant μ was 0.42 ± 0.02 h⁻¹, which was ~1.7-fold lower compared to wt. Such a role of Hyd-3 and Hyd-4 enzymes might be interpreted if Hyd-3 and Hyd-4 are contributing toward H₂ cycling via Hyd enzymes and proton transport through proton ATPase (Trchounian and Sawers 2014) and their absence disturbs the H₂ cycling and proton transport and thus reduces μ . Note that similar reduction of μ in *hycE* mutant during fermentation of glucose, glycerol and formate was experimentally shown at pH 5.5 (Mirzoyan et al. 2018). Further study is required to elucidate the exact role of Hyd enzymes depending on the presence of carbon sources mixtures or organic wastes and change of external parameters.

H₂ production during utilization of SCG by *E. coli*

Escherichia coli wt produced 30.7 mL H₂ (g sugar)⁻¹ (P < 0.03) starting at 24 h pending not longer than 48 h with the rate of - 1.3 mL H₂ (g sugar)⁻¹ h⁻¹. To reveal how much H₂ was produced per kg

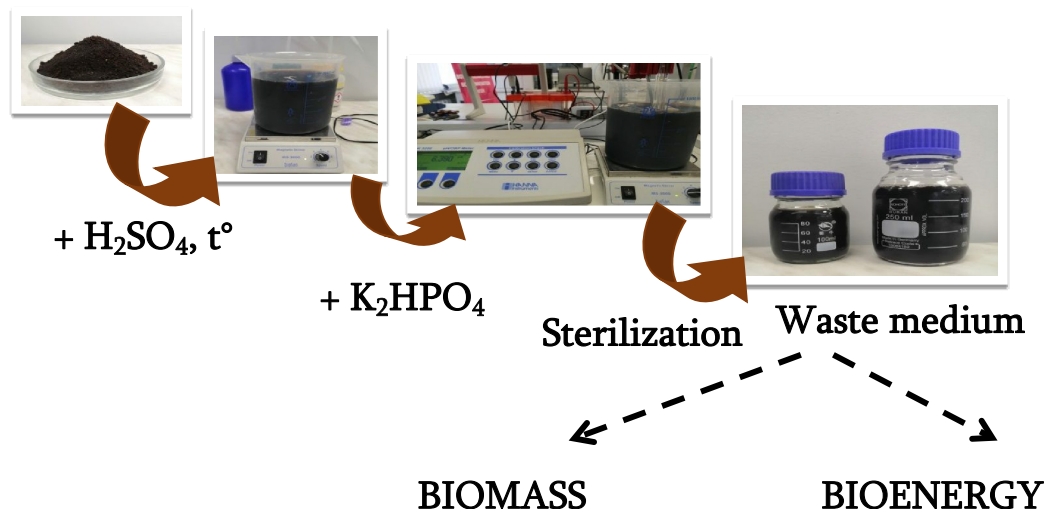


Figure 1. Pretreatment steps of SCG for biomass and bioenergy production. For details, see the Materials and methods section.

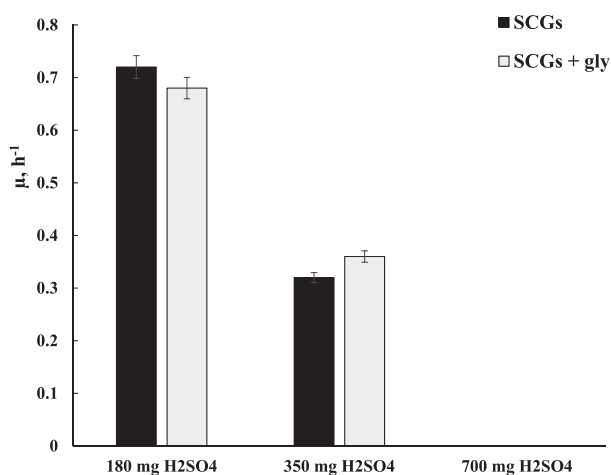


Figure 2. Specific growth rate of *E. coli* BW25113 wild type during utilization of 40 g L⁻¹ of SCGs treated with 180 mg H₂SO₄ for 45 min, 350 mg H₂SO₄ for 20 min and 700 mg H₂SO₄ for 20 min, where H₂SO₄ concentrations were calculated for 1 g SCGs; gly is glycerol.

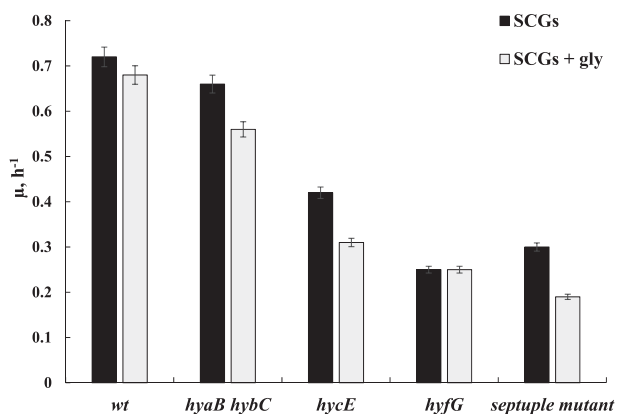


Figure 3. Specific growth rate of *E. coli* BW 25113 wild type (wt) and mutants during utilization of SCG. Septuple mutant: *hyaB hybC hycA fdoG ldhA frdC aceE* (see Table 1); for other details, see the Materials and methods section and legends to Fig. 2.

SCG used, it was calculated that *E. coli* wt cells yielded 2.75 L H₂ (kg SCG)⁻¹.

To understand which Hyd enzyme is responsible for H₂ production during utilization of SCG, different Hyd mutants (see Table 1) have been applied. This was important since Hyd enzymes are reversible and can work in different directions under various conditions including substrates (Mnatsakanyan, Bagramyan and Trchounian 2004; Trchounian 2015; Trchounian, Sawers and Trchounian 2017). Actually, similar maximum yields were observed in mutants used, besides *hycE* where H₂ production was absent. These data clearly showed that during anaerobic utilization of SCG Hyd-3 is the main Hyd enzyme responsible for H₂ production, which was demonstrated for sole glucose fermentation (Maeda et al. 2018; Petrosyan et al. 2020). In *hyfG* mutant, H₂ production was determined till 48 h, as in wt (Fig. 4). Addition of glycerol to SCG did not enhance H₂ production that has similar yields and rates as in SCG utilization only. It should be noted that various organic wastes and different bacteria have been applied to produce H₂ when different yields and rates have been obtained (Kapdan and Kargi 2006; Kumari and Singh 2018). Particularly, mixture of inoculum from anaerobic sludge that was applied has been reported to yield 56.47 mL H₂ (g VS)⁻¹ when using mixture of corn stalk, bean husk and municipal solid waste (Sekoai and Kana 2013). Recently, it was also shown that from food waste by anaerobic mixed cultures it was possible to enhance the H₂ production yield till 71.34 mL (g VS)⁻¹ (Rafieenia, Pivato and Lavagnolo 2019).

Furthermore, in septuple mutant, during the utilization of SCGs hydrolysate H₂ production yield was enhanced ~2-fold and reached 72 mL H₂ (g sugar)⁻¹ or 5.5 L H₂ (kg SCG)⁻¹ ($P < 0.03$) with H₂ production rate of ~3 mL H₂ (g sugar)⁻¹ h⁻¹ at 48 h of growth. The H₂ yields observed in septuple mutant are higher compared to different bacteria and conditions, where mixed cultures were used. These data seem to be very promising as it will be possible to further increase the H₂ yield by optimizing the pretreatment methods and external parameters. Indeed, almost similar maximum H₂ yields were observed during SCGs and glycerol co-fermentation. The use of mixture with glycerol did not enhance H₂ generation in septuple mutant, as Hyd-1 and Hyd-2 were responsible for H₂ generation during glycerol fermentation (Trchounian and Trchounian 2009). It can be suggested that the

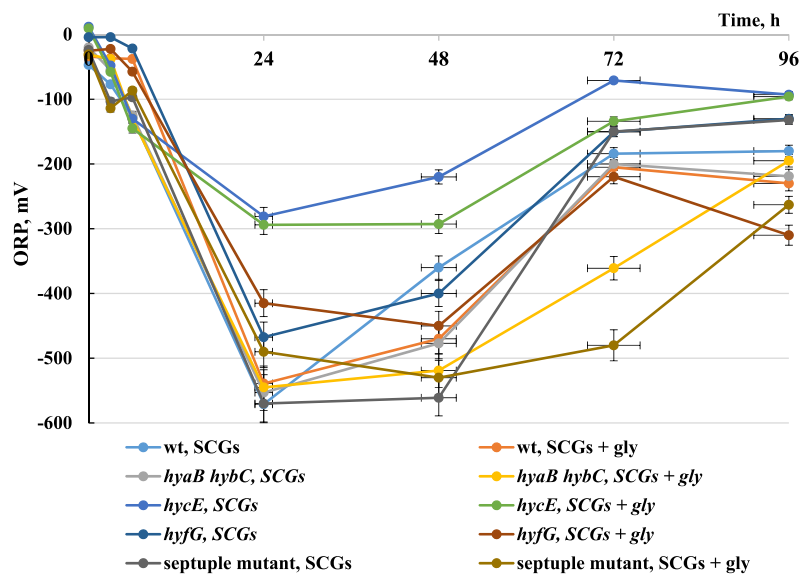


Figure 4. ORP and H₂ production during growth of *E. coli*. Bacteria were grown upon utilization of 40 g L⁻¹ SCGs or SCGs + glycerol (SCGs + gly) where indicated for 96 h at 37°C. For more details, see the Materials and methods section.

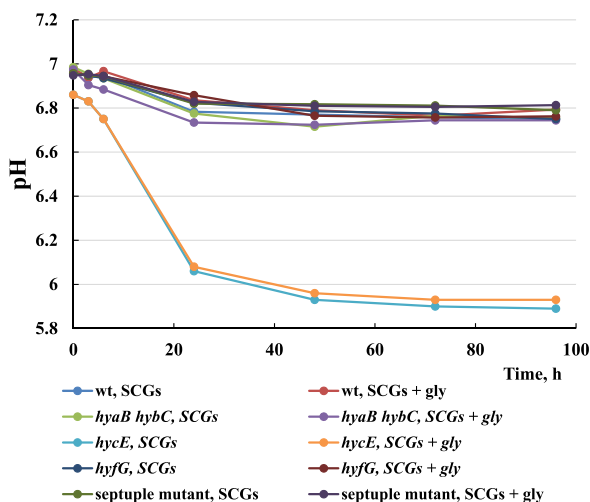


Figure 5. External pH during growth of *E. coli*. For more details, see the legends to Fig. 4.

engineered septuple mutant can be applied when sugars are utilized. The results are in good conformity with the initial strategy to construct the septuple mutant for enhanced H₂ production from sugars (Maeda, Sanchez-Torres and Wood 2007b). Further study is required to optimize the pretreatment methods and test them during scale-up.

The external pH in *hycE* mutant was dropped significantly by 0.9 units till pH of 5.9 compared to wt, and in *hyaB hybC* double and *hyfG* single mutants where the medium was acidified by 0.2–0.3 units till pH of 6.7 (Fig. 5). As it was shown, Hyd-3 was responsible for H₂ generation during fermentation and mainly formate, as one of the end products is formed and secreted contributing to the acidification of the external pH, whereas in *hycE* mutant the formate cannot be oxidized to H₂ and CO₂ or alternatively it can affect the secretion of formate. In addition, Hyd enzymes

contribute to proton motive force generation during fermentation (Trchounian, Blbulyan and Trchounian 2013), and secretion of organic acids can generate pH gradient and/or membrane potential (Michels et al. 1979). It is suggested that one of the mechanisms of intra- and extracellular formate concentration regulation is related with maintenance of proton motive force. Such idea about the role of Hyd enzymes in addition to the existing one for organic acids had been reviewed recently (Trchounian and Trchounian 2019). The role of different Hyd enzymes on the generation of formate gradient and other organic acids during fermentation needs separate and more comprehensive study.

It is important to mention that the main goal of the research was focused on evaluating the possibility of utilization of SCG for biomass and biohydrogen production by *E. coli* wild type and further application of the engineered strain/s for enhancement of H₂ generation in the waste used. In addition, the obtained results were promising; thus, the next step will be the determination of fermentation end products that will add new valuable information for regulating cell growth and biohydrogen production.

CONCLUSIONS

Taken together, the results obtained have shown that SCGs hydrolyzed with H₂SO₄ for 45 min and with 2 times dilution were optimal conditions for *E. coli* to grow and produce H₂ with yield of ~31 mL H₂ (g sugar)⁻¹ or 2.75 L (kg SCG)⁻¹. Hyd-3 and Hyd-4 had influenced specific growth rate during utilization of SCG, which might be due to change of bioenergetic properties. *Escherichia coli* Hyd-3 was responsible for H₂ production, where deletion of Hyd-1 and Hyd-2 had no significant difference. H₂ production yield and rate were enhanced by ~2-fold in septuple mutant reaching ~72 mL H₂ (g sugar)⁻¹ or 5.5 L H₂ (kg SCG)⁻¹.

Bioconversion of organic wastes like SCGs that are generated in huge amounts, and the utilization of these sources for production of added-value products such as bio-H₂, is of great interest. The data can be used for further technological

optimization of coffee wastes as a source for not only biomethane and bioethanol but also biohydrogen production.

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Conflict of interest. None declared.

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